

# Model-assisted engineering of *Escherichia coli* for biofuel production

David Fell, Mark G Poolman, Hassan Hartman  
Oxford Brookes University

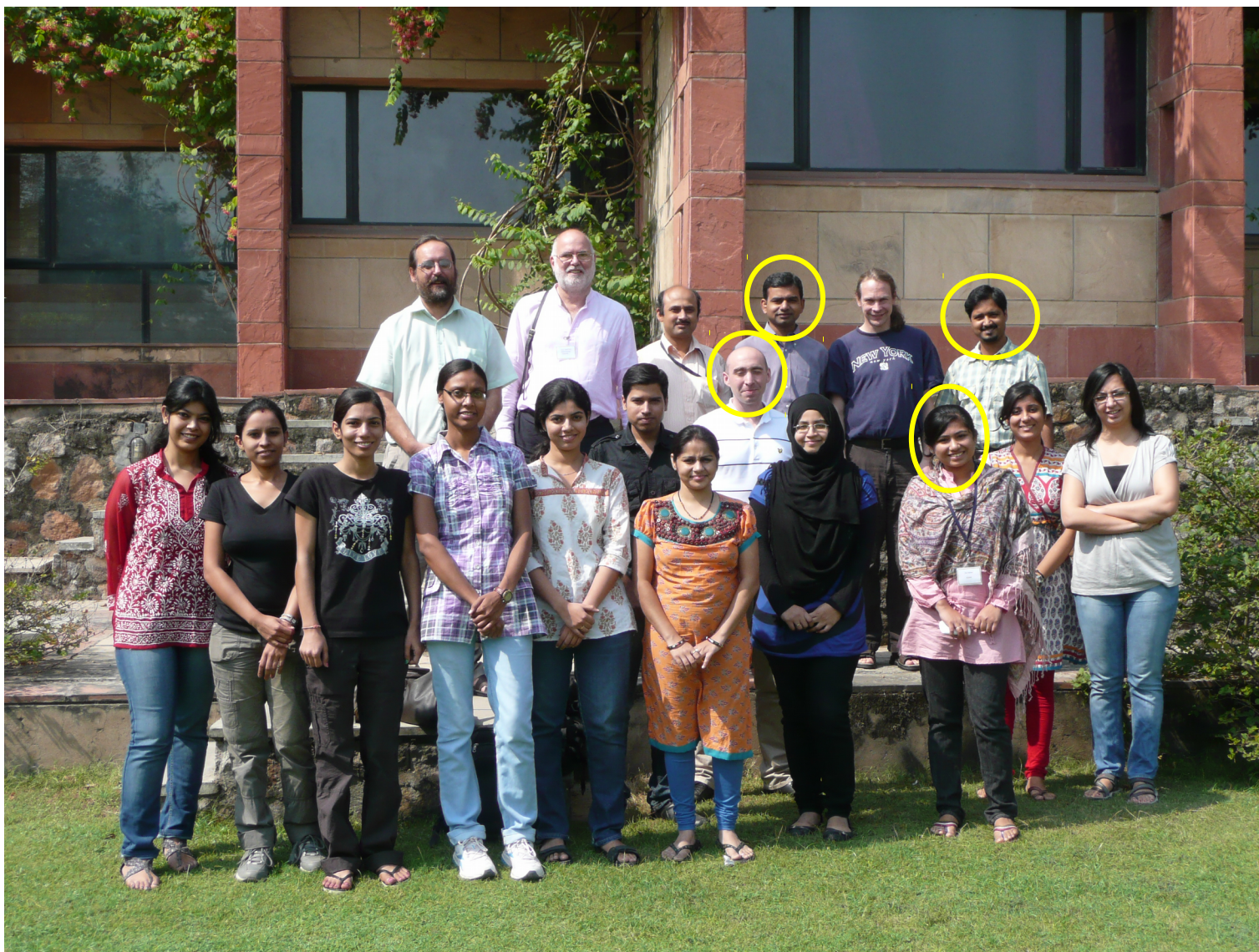
Zia Fatma, Shireesh Srivastava, Syed Shams  
Yazdani  
ICGEB, New Delhi

OXFORD  
BROOKES  
UNIVERSITY



Department of Biotechnology  
Govt. of India

# ICGEB Delhi 2011



# Biofuels

Ethanol	Corn or potato starch, sugar cane	Lignocellulosic waste
Butanol	Corn starch	Lignocellulosic and food waste
Biodiesel (alkanes, esterified fatty acids)	Palm oil, oil seeds	Photosynthesis; waste cooking oil; ? lignocellulosic waste





Pilot plant for production of ethanol from lignocellulosic waste, Kashipur, India. Designed by ICT Mumbai Centre for Energy Biosciences. Photo: courtesy of Prof A  
June 2018



# Microbial Routes to Biodiesel

- Cyanobacteria naturally produce alkanes and long-chain fatty alcohols by photosynthesis.
- Some algae and yeasts produce high levels of triglycerides as storage compounds, which could be trans-esterified to fatty methyl esters.
- Engineering of bacteria or yeasts with the cyanobacterial pathway to generate alkanes non-photosynthetically from C-containing substrates

# Approach in this Study

- Used *E coli* as the host organism because of:
  - Detailed knowledge of its metabolic network
  - Extensive molecular genetics tools that facilitate experimental modification of metabolism
  - Grows readily
  - Native ability to use pentoses ( such as xylose from lignocellulose wastes) in addition to glucose



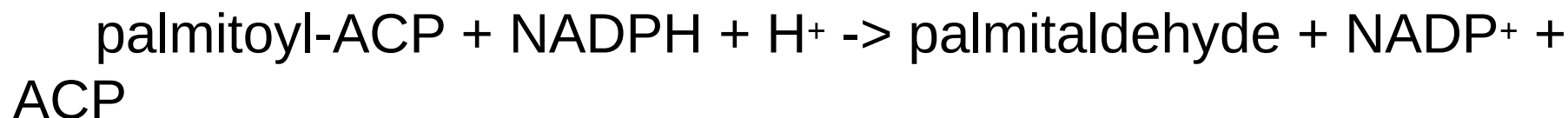
# Heterologous Alkane Synthesis Pathway

- The pathway branches off from fatty acid synthesis by intercepting acyl-ACP intermediates.
- AAR, acyl-ACP reductase, EC 1.2.1.80 from *Synechococcus elongatus* releases a long-chain aldehyde, e.g.:  
$$\text{palmitoyl-ACP} + \text{NADPH} + \text{H}^+ \rightarrow \text{palmitaldehyde} + \text{NADP}^+ + \text{ACP}$$
- ADO, aldehyde oxygenase deformylating, EC 4.1.99.5 from *Nostoc punctiforme* :  
$$\text{long-chain aldehyde} + \text{O}_2 + 2\text{NADPH} + 2\text{H}^+ \rightarrow \text{alkane} + \text{formate} + \text{H}_2\text{O} + 2\text{NADP}^+$$

The reaction requires ferredoxin and ferredoxin reductase

# Alternative to Long-Chain Alcohols

- AAR, acyl-ACP reductase, EC 1.2.1.80 from *Synechococcus elongatus* releases a long-chain aldehyde, e.g.:



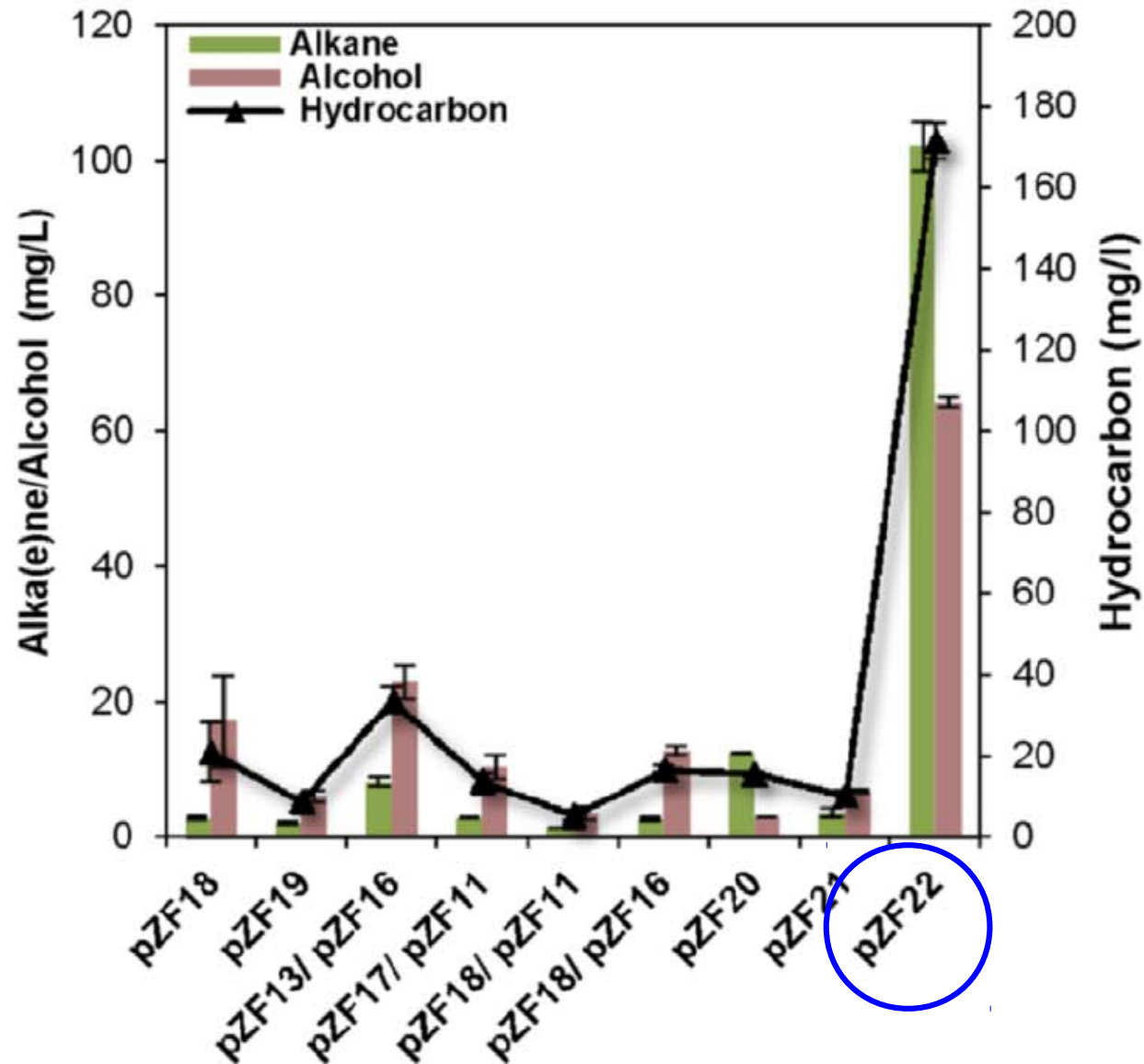
- YbbO, NADP<sup>+</sup>-dependent aldehyde reductase, EC 1.1.1.2 from *E. coli*, overexpressed:  
long-chain aldehyde + NADPH + H<sup>+</sup> → long-chain alcohol + NADP<sup>+</sup>



# Optimisation at ICGEB

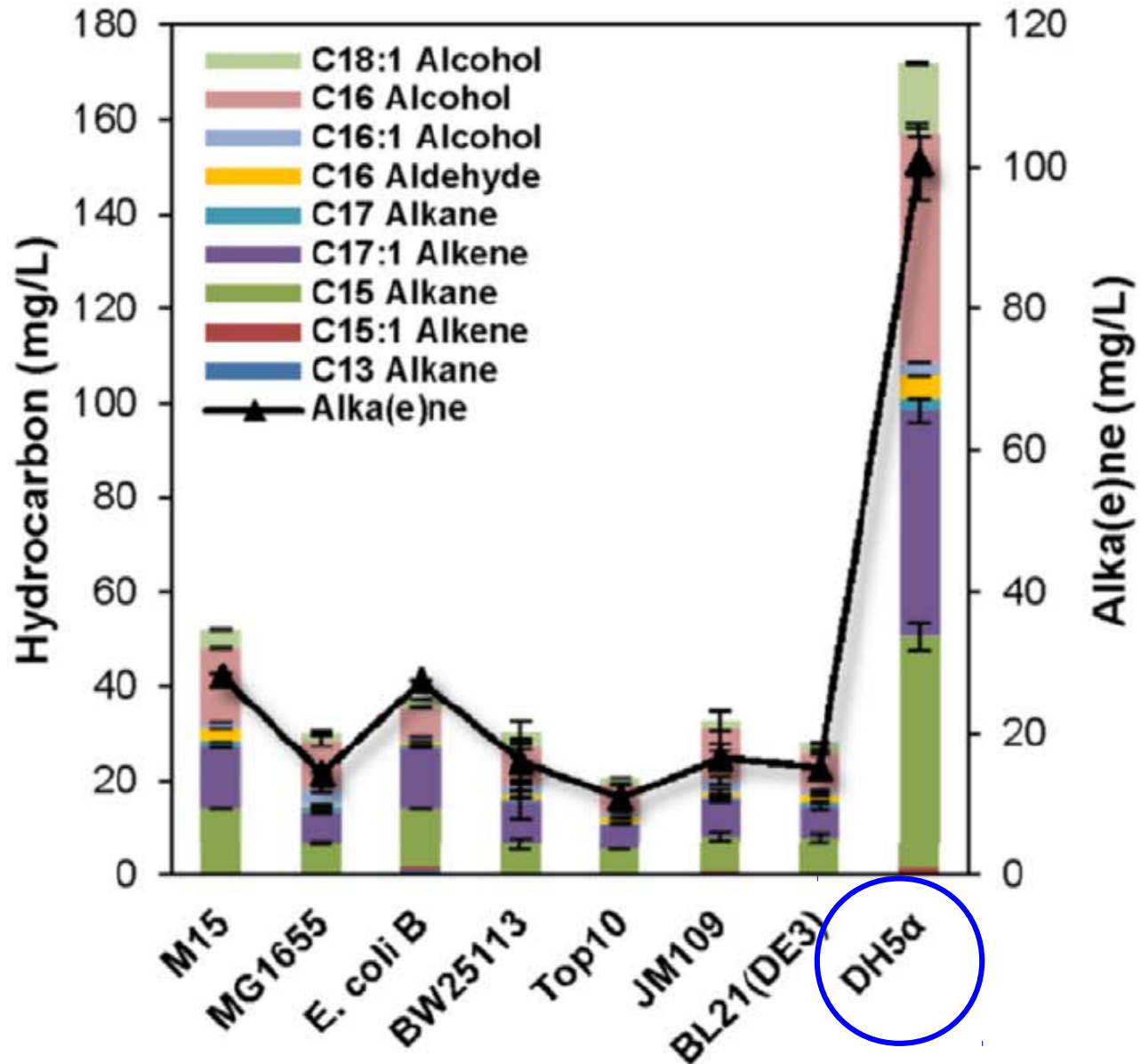
- Different options of enzyme sources (after codon optimisation) and promoters were explored via expression in medium-copy number plasmids.
- A fusion protein of AAR and ADO was tested but not found better than enhanced expression of separate genes with T5 promoters.
- For fatty alcohols, expression of AAR and YbbO from T5 promoters was found better than expression of AAR alone and reliance on native activity of *E coli* alcohol dehydrogenases.

# Outcome for Alkanes





# Outcome for Strain Selection



# Modelling for Improved Productivity



# Metabolic Modelling Methods

## Structural modelling techniques

- - need an accurate reaction list from which to generate a stoichiometry matrix; assume metabolic steady state.
- - show existence (and number) of feasible metabolic routes; optimal conversion stoichiometries; network flux values.

# Metabolic Modelling Methods

## Kinetic modelling techniques

- - need a reaction list **and** full kinetic description of each enzyme/step.
- - predict time-courses, steady-state values of reaction fluxes **and** metabolite concentrations.
- - allows sensitivity analysis (Metabolic Control Analysis) to compute dependence of fluxes and concentrations on enzyme activities.

# Structural Modelling Methods

## Elementary modes analysis

- - all feasible routes (modes) through a network from nutrients to metabolic products;
- - network flux values and product yields;
- - good for designing knock-out strategies to eliminate metabolism to unwanted products;
- - computationally limited to small to medium sized metabolic networks.

(Schuster, Dandekar & Fell, 1999, 2000)

# Structural Modelling Techniques

## Linear programming (LP or Flux Balance Analysis)

- - incorporates known metabolic properties, such as nutrient uptake rates, as constraints;
- - computationally feasible even on the largest (genome-scale) metabolic models;
- - can be used to design over-expression strategies for increasing productivity;
- - determines the optimal network route to achieve a specified metabolic objective;
- - several techniques for design of knock-out strategies, though very large models produce less clear results;
- Basic method only produces a single solution; finding multiple optima or near-optimal solutions is more complex.

(Fell & Small, 1985; Varma & Palsson, 1993)



# Our Modelling Approach

Though an elementary modes analysis would potentially have been feasible:

- As alkanes are not catabolic products, merely cutting our routes to other products would not necessarily induce alkane synthesis, and
- We expected to have to over-express parts of central carbon metabolism to supply enough substrate to allow significantly increased flux through the fatty acid synthesis pathway.

We therefore opted for a linear programming approach.

# The Model

- The model was based on a central carbon metabolism (CCM) model developed by Trinh, Unrean & Sreenc (2008).
- It was reconstructed using *ScrumPy* from the EcoCyc database, plus the additional heterologous reactions to alkane.
- A single, representative alkane – pentadecane – was modelled as output; the carbon source was glucose in aerobic conditions.
- Growth of the cell was modelled by withdrawal of a small set of CCM intermediates at appropriate rates established from larger-scale *E coli* models.
- The model has 74 reactions and 61 metabolites.



# Initial Model Analysis

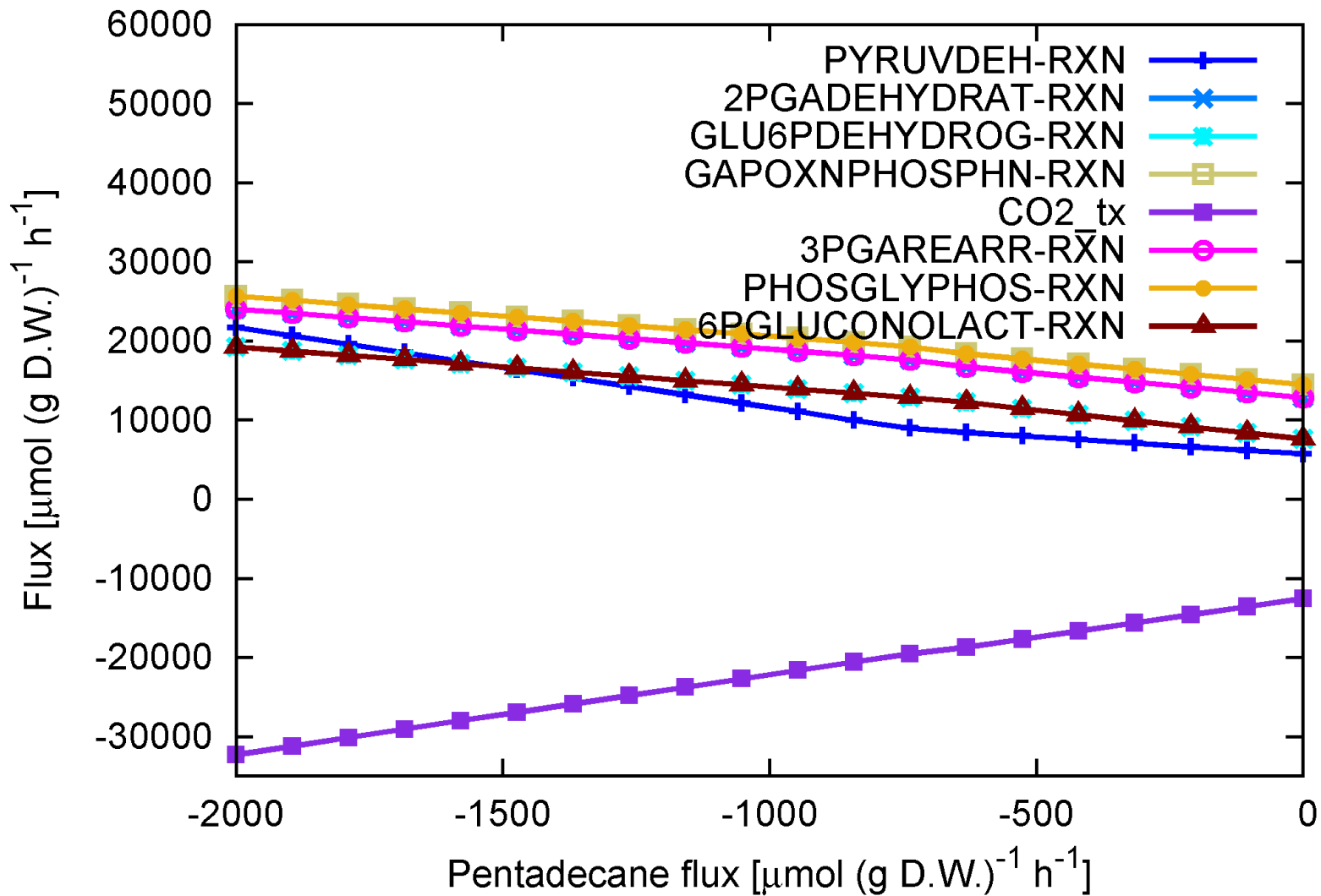
1. The model was checked for stoichiometric and energetic consistency.
2. LP was used (with *ScrumPy*) to check the ability to supply biomass precursors from glucose at a rate equivalent to a growth rate of  $1 \text{ g dcw.h}^{-1}$  (as a constraint), with minimisation of total flux in the network as the optimisation criterion.
3. The rate of glucose uptake was then set as a constraint at twice its value in the previous solution to model the fate of excess carbon intake. Only acetate and lactate were formed in addition; no pentadecane.



# Constraint Scanning

- The model can be solved for simultaneous production of biomass and pentadecane by imposing these as constraints.
- However, if we use ScrumPy to compute a series of LP solutions for fixed biomass but pentadecane from 0 to 2 mmol.(gDW.h)<sup>-1</sup>, we can see how fluxes through the network have to change to support alkane synthesis.

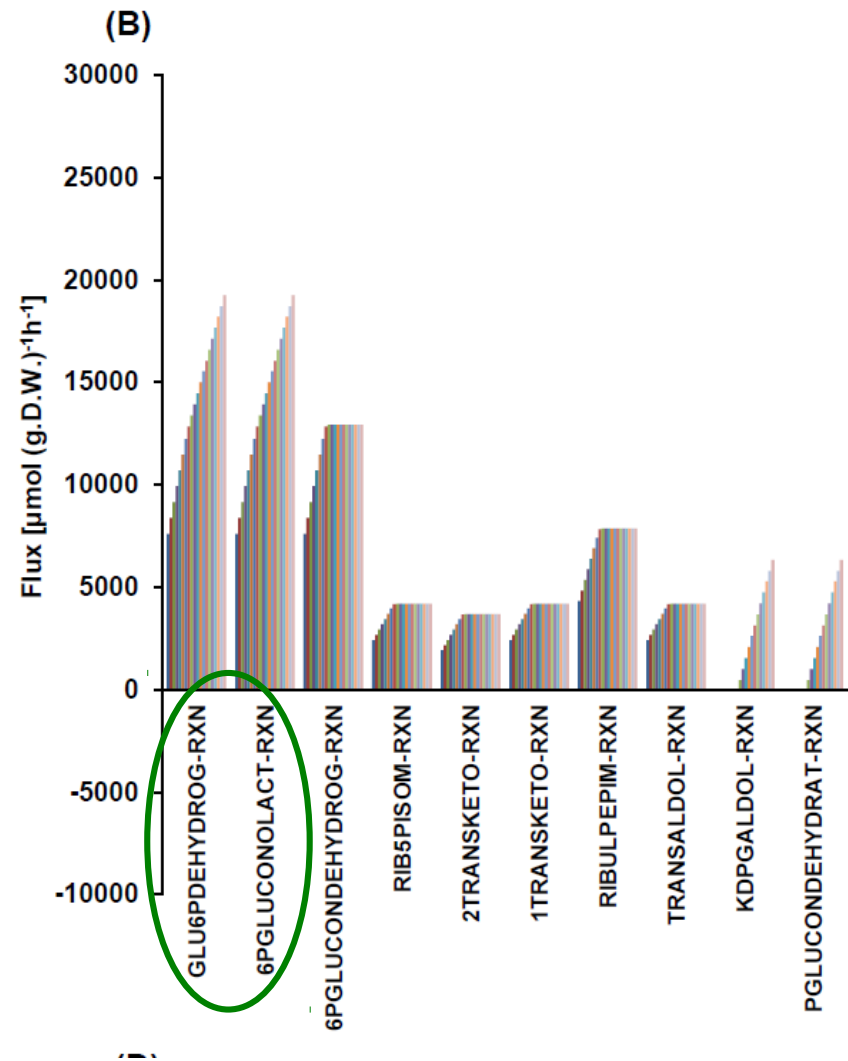
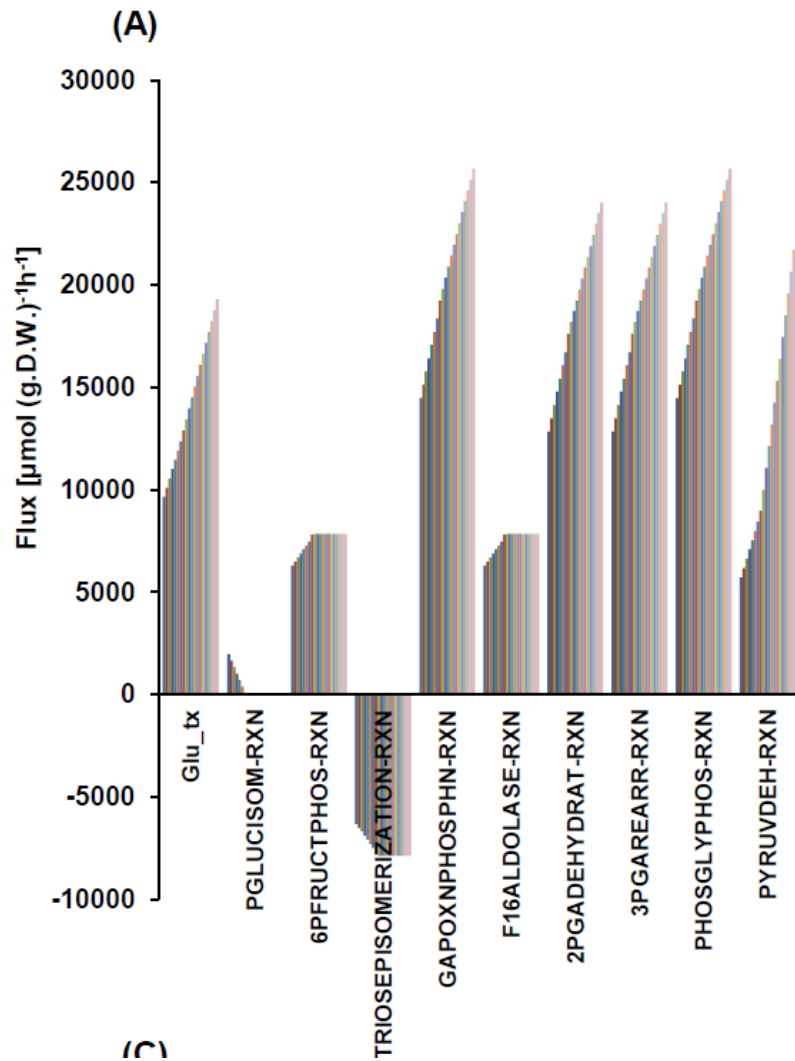
# Constraint Scanning Alkane Production



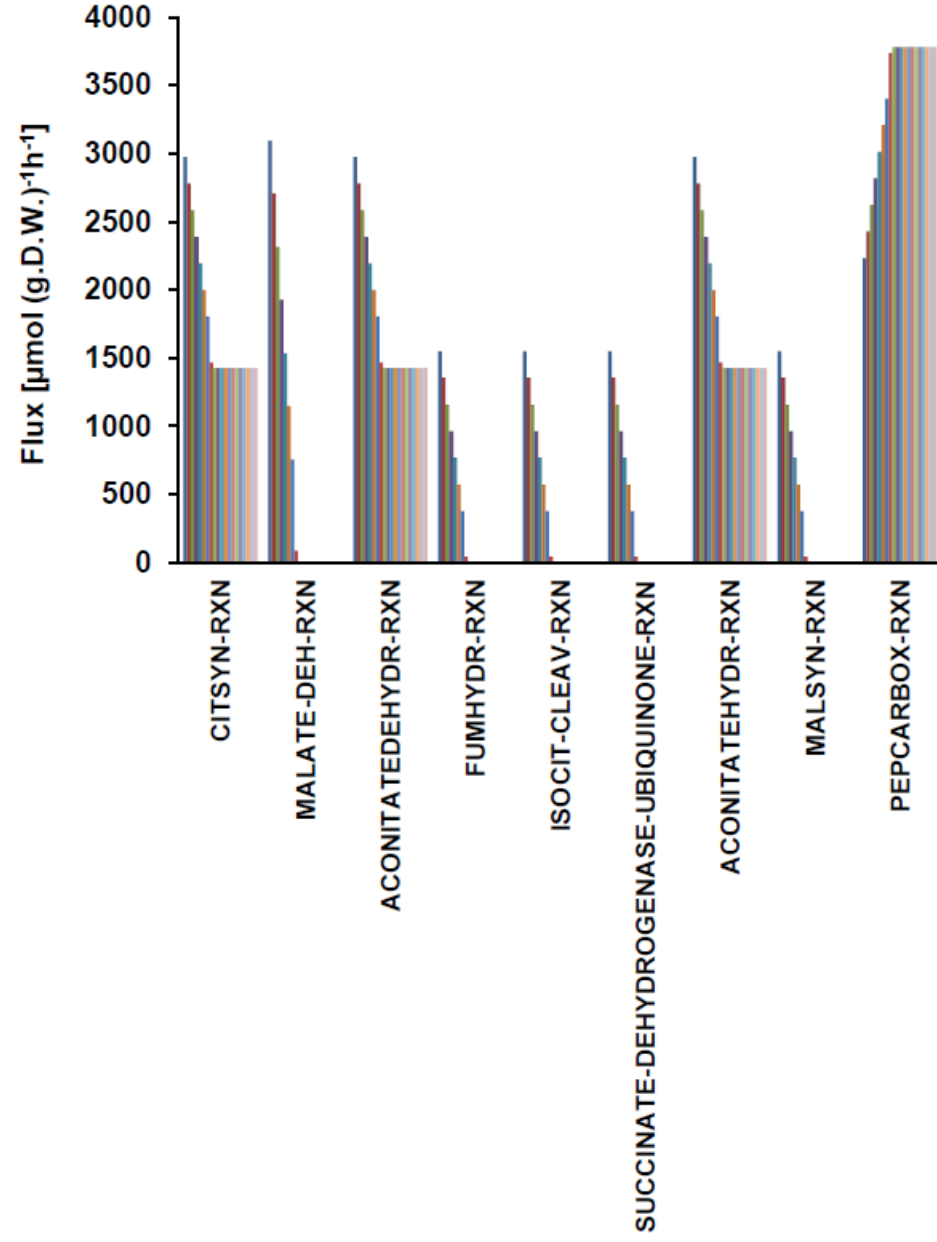
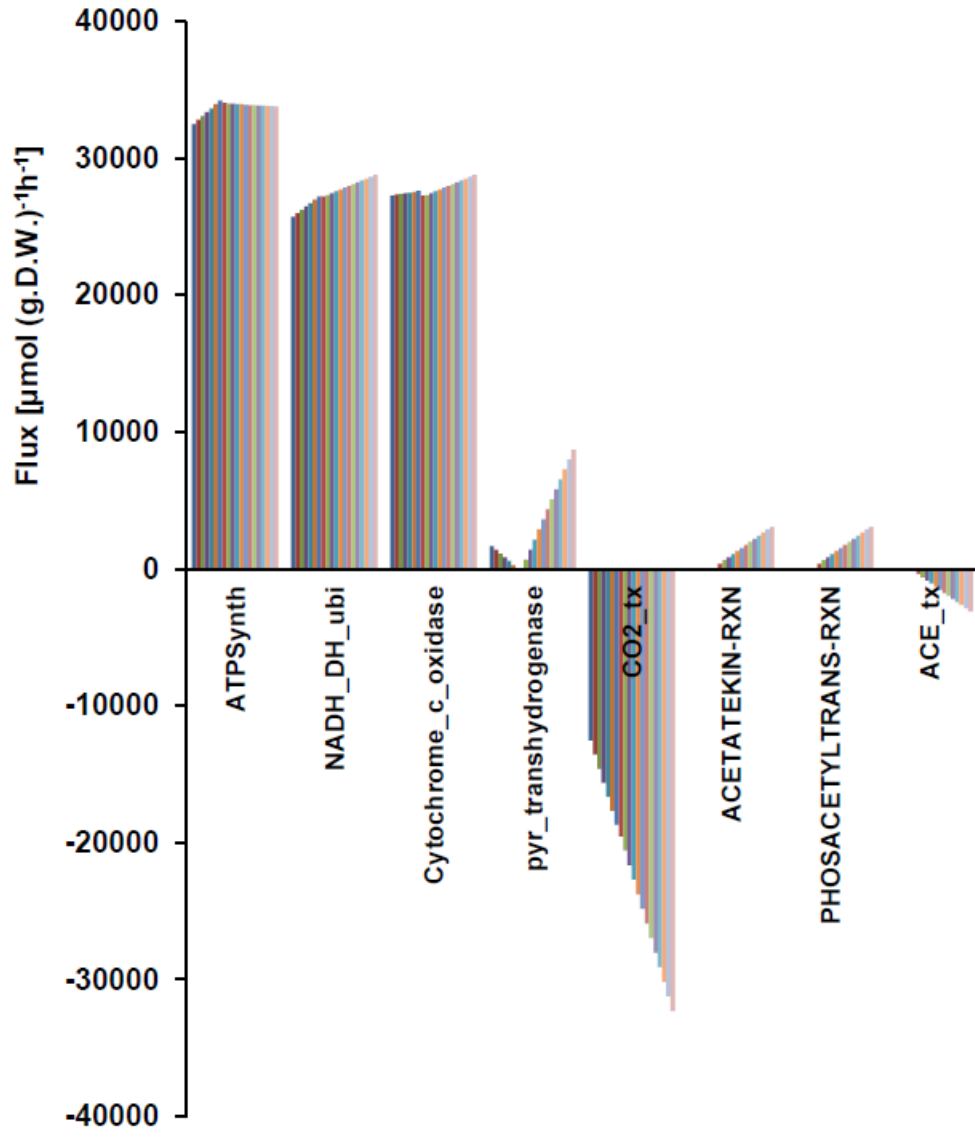
# Reaction Correlations



# Another View ...



... continued





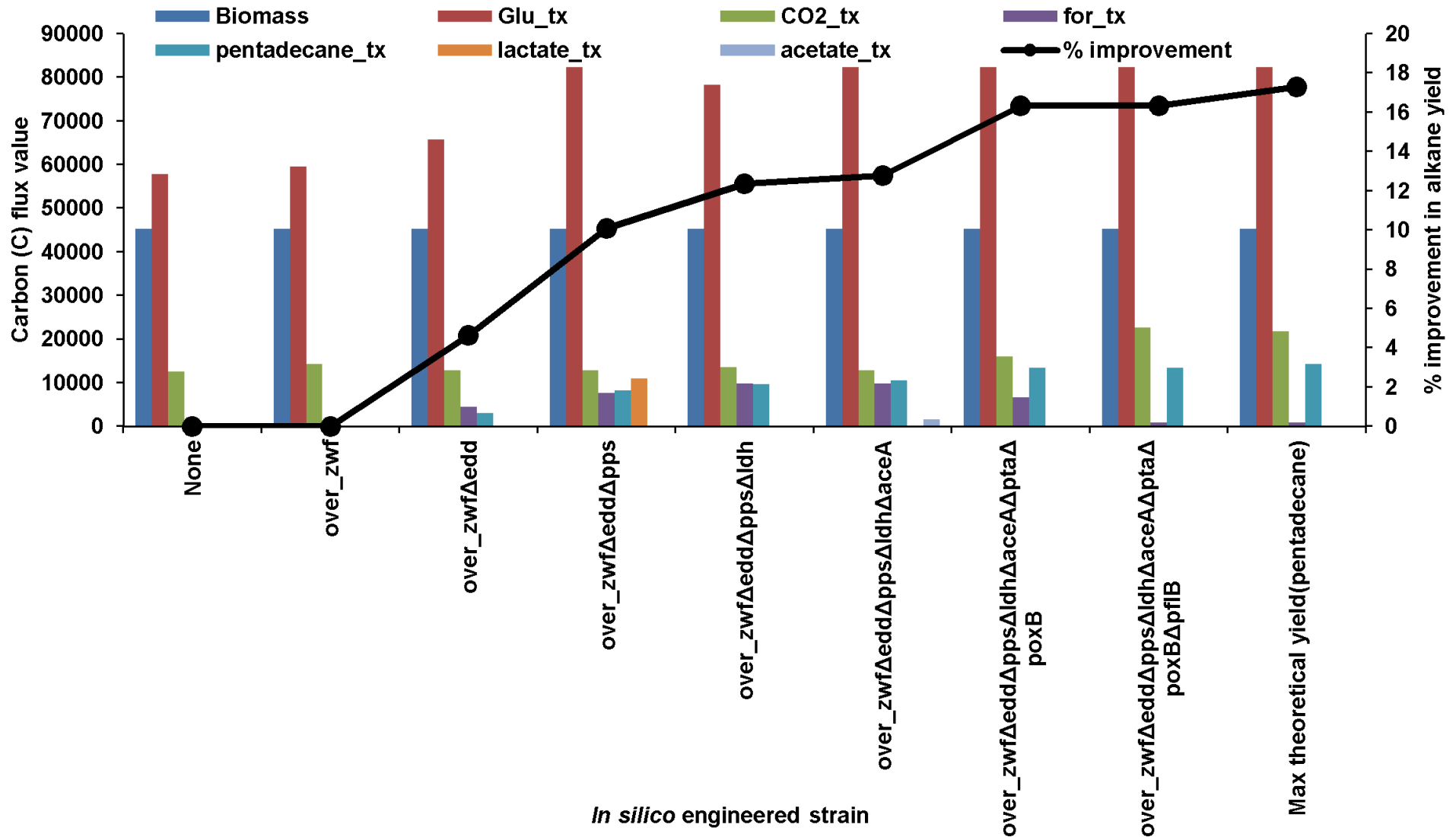
# G6PDH Over-Expression

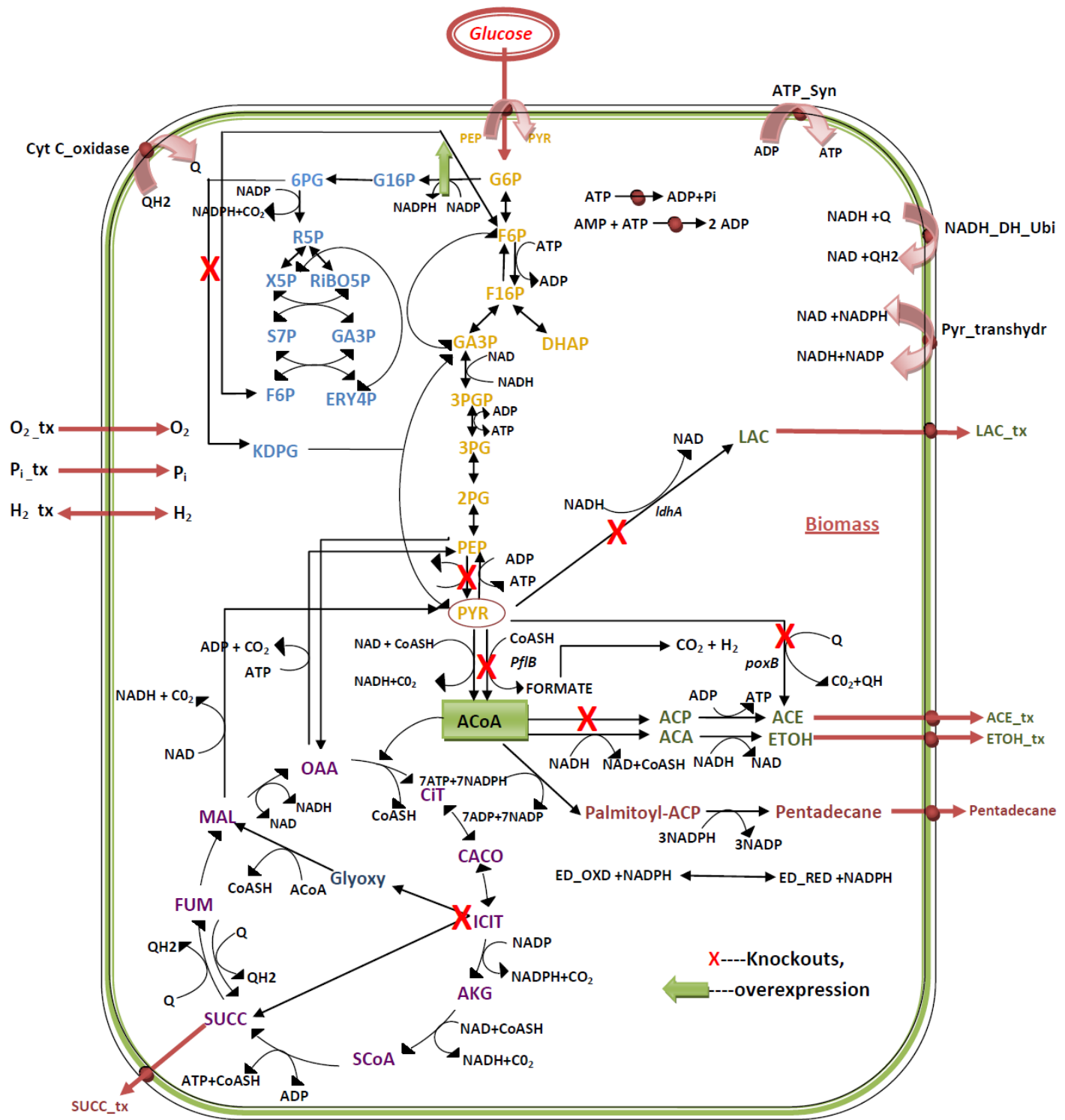
- On the basis that large relative increases in flux with pentadecane production could indicate steps that might become limiting, glucose-6-phosphate dehydrogenase (G6PDH) was selected as a target for over-expression.
- This was represented in the model by setting its flux to twice the value needed for biomass formation. Solving the model with this as a constraint only resulted in extra flux being diverted into the Entner-Duodoroff pathway, and then from pyruvate to PEP via PEP synthase with no change in other outputs apart from CO<sub>2</sub>.

# Knock-out Selection

- As the increased C-flux was not reaching alkane, we modelled a knockout of the ED pathway by constraining the flux of the reaction catalysed by the *edd* gene product to zero, resulting in small amounts of formate and alkane.
- This was followed by deleting the PEP synthase reaction ( $\Delta pps$ ), which resulted in larger amounts of formate, alkane and lactate.
- As additional products appeared, we modelled additional knock-outs in the branches leading to them until the only products were biomass, CO<sub>2</sub> and alkane.

# Knock-outs for Alkane production

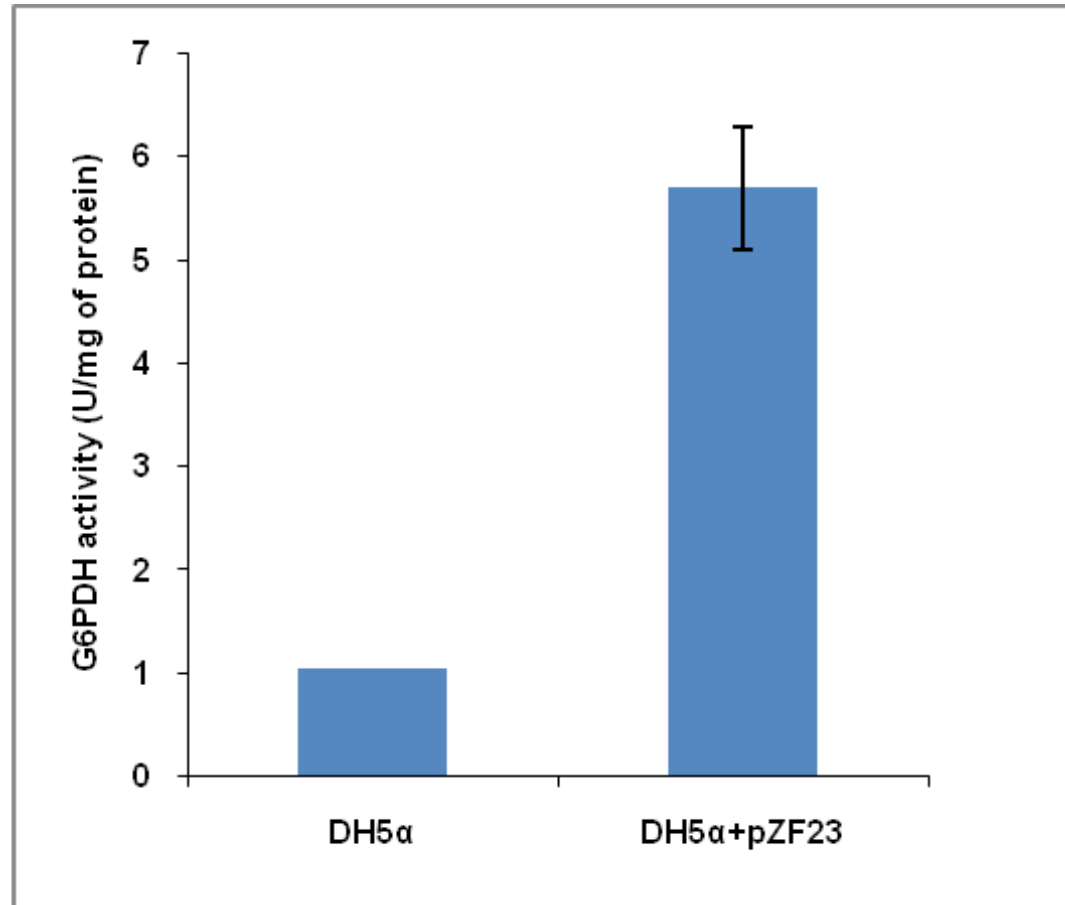




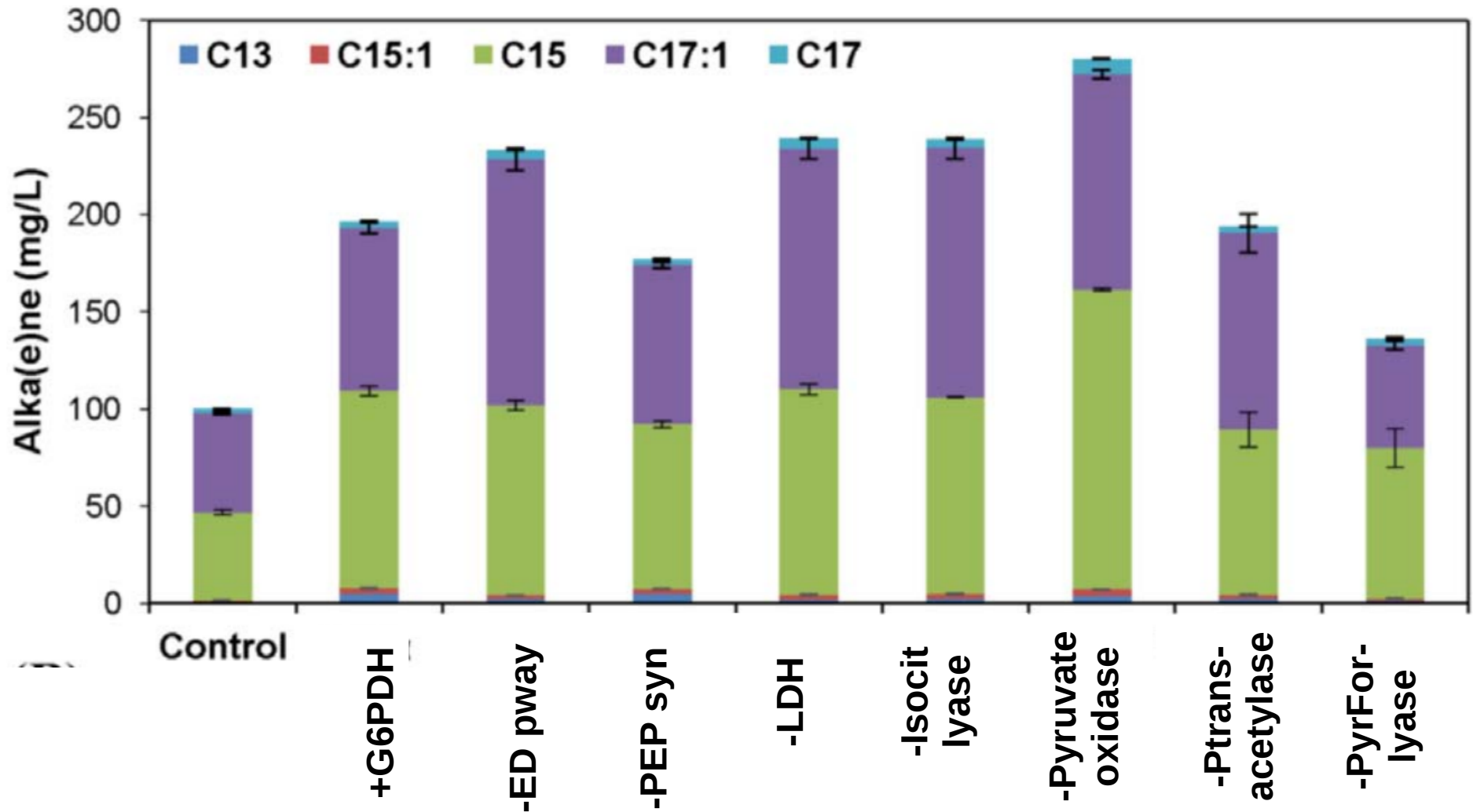
# Experimental Implementation



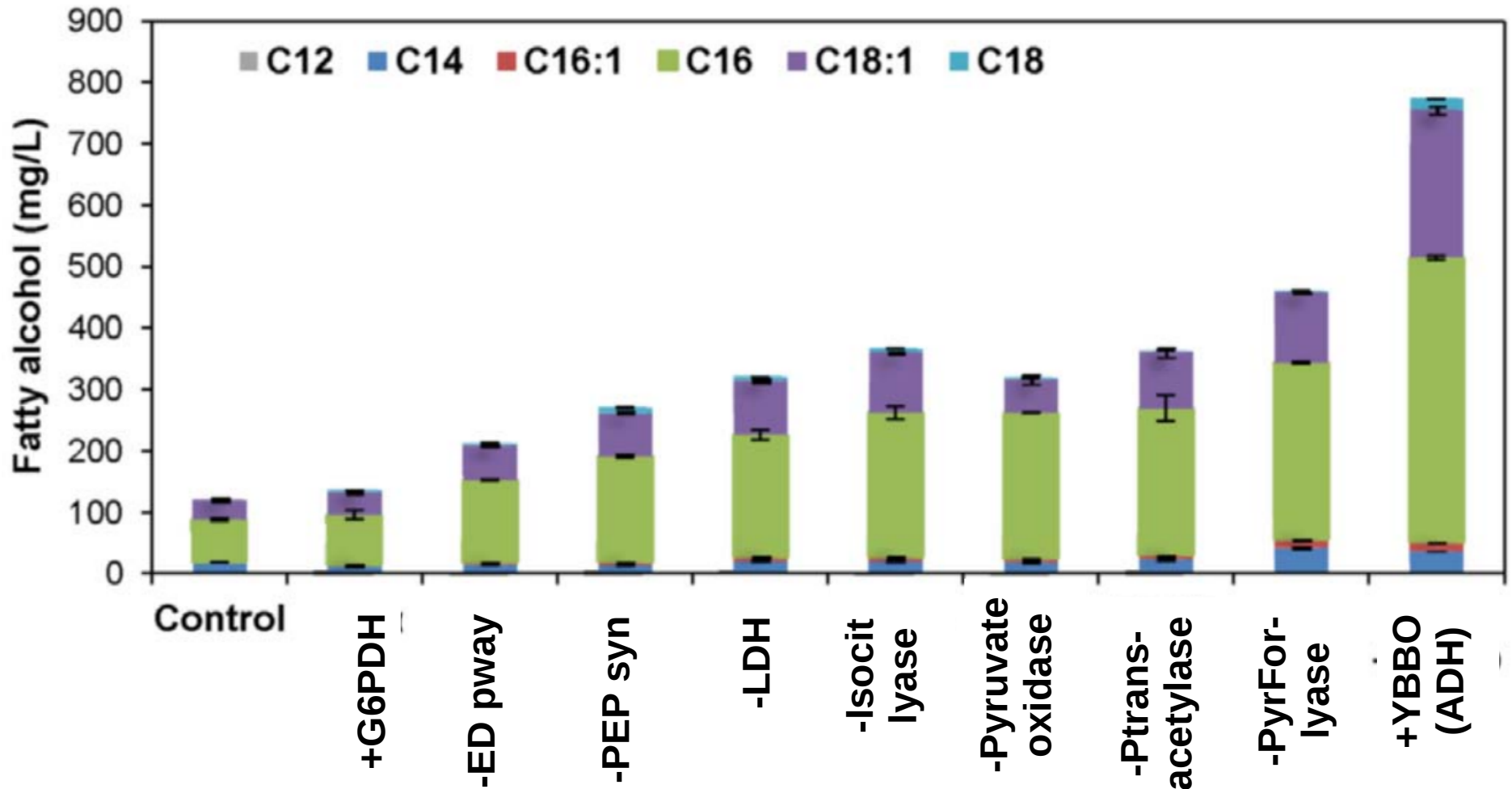
# G6PDH Over-expression



# Alkane Yields



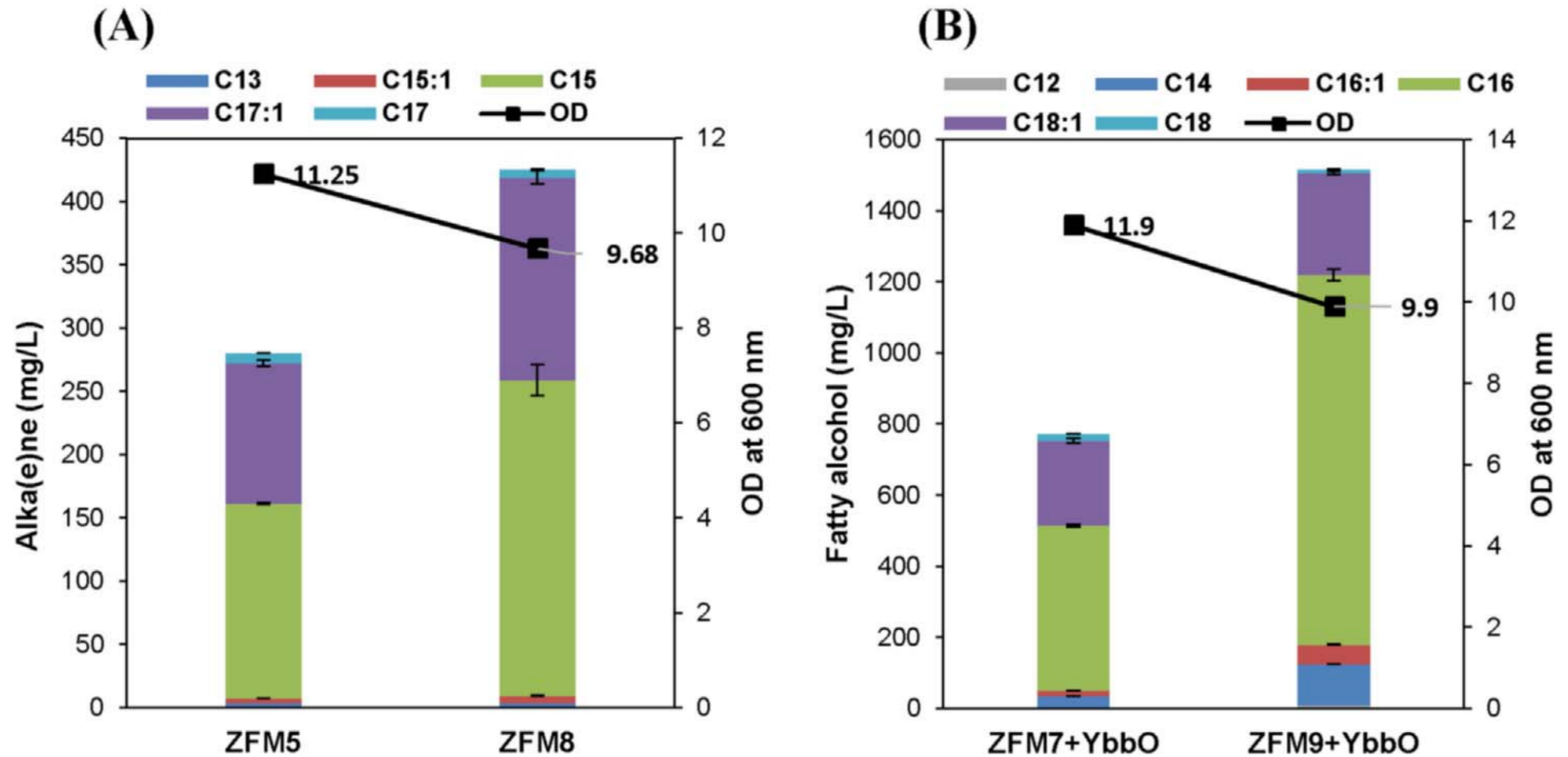
# Fatty Alcohol Production



# Final Improvement

- In the final strain so far, the carbon yield from glucose was still less than the theoretical maximum attainable with simultaneous biomass synthesis.
- Analysis of the flux values in the model solutions showed that the main competitive flux for alkane was now the phospholipid synthesis for biomass.
- This was therefore attenuated by a non-lethal knockout of a phospholipid pathway gene for the final producing strain.

# Increase in Yield by Growth Attenuation



# Overall Result

- The initial strain optimisation by ICGEB improved alkane production from an initial 2.8 mg/L to 102 mg/L.
- The further model-designed improvements led to a further increase to 425 mg/L.
- The productivity of fatty alcohols was increased even further to 1500 mg/L.
- These strains, when tested in fed-batch bioreactors produced 2.54 g/L alkane and 12.5 g/L fatty alcohol – the highest alkane yields yet reported for *E coli*.

# Conclusions

- Theory-led design can narrow the search space for metabolic engineering.
- There is still a long way to go to achieve commercial alkane production by microorganisms from sustainable C-sources.
- Fatma, Z. et al Model-assisted metabolic engineering of *Escherichia coli* for long chain alkane and alcohol production, *Metabolic Engineering*, **45**, 134-141 (2018).



# Growth curves

